## ENVIRONMENTAL ASSESSMENT ON THE AQABA GULF COASTAL WATERS; EGYPT

## AHMED M. ABDEL-HALIM, ABOEL-KHAIR, E. M, FAHMY M. A, Shridah M. A.

National Institute of Oceanography and Fisheries, Anfoushy, Alexandria, Egypt. E-mail: aabdel\_halim@hotmail.com

Key words: Aqaba Gulf – Coastal water – nutrient salts – hydrographic parameters – Trophic state.

#### ABSTRACT

Environmental Information and Monitoring Programme (EIMP) for the Egyptian coastal waters of Aqaba Gulf was established to initial monitoring and data base system by applying quality control assessments in order to evaluate, protect and sustainable use of coastal regions. Within the framework of this programme six bimonthly field campaigns were carried out annually during 1998 - 2004. A total of 11 coastal stations were selected to cover different locations of the Gulf. The surface distribution patterns of hydrographical parameters (water temperature, salinity, dissolved oxygen and pH) as well as chlorophyll-a, total suspended matter, transparency, nitrogen and phosphorus forms and reactive silicate were investigated. The obtained data of seven years work deduced that no thermocline or thermal pollution, variations in the pH and salinity values were insignificant and the water was well-oxygenated. The concentrations of nitrogen and phosphorus in the dissolved and total forms as well as reactive silicate were found quite low. The abundance of inorganic nitrogen forms were found in the order  $NH_4-N > NO_3-N \ge NO_2-N$  reflecting the increasing rate of NH<sub>4</sub>-N production than its uptake rate as compared with the other inorganic nitrogen forms. The study showed that, nitrogen and phosphorus were found in Agaba Gulf coastal waters, principally, in organic forms. The low levels of obtained during the present away for each of Chl-a, TSM, NH<sub>4</sub>-N, NO<sub>3</sub>-N, TN, PO<sub>4</sub>-P and TP signified that Aqaba Gulf Coastal waters can be classified between oligotrophes to mesotrophic state. Accordingly, it is safe to conclude that the area of study is not yet seriously threatened, in spite of the rapid recreational and human developments taken place on its coast during the last ten years.

#### **1. INTRODUCTION**

The Red Sea, with a length of about 1930 Km and average width of 280 Km, lies between 12° 30' N and 32° 44' E. The Red Sea joins Gulf of Aqaba and Gulf of Suez in the northern part. The Sinai Peninsula divides the Red Sea into the shallow Gulf of Suez (55 - 73m average depth) and the deep Gulf of Aqaba (average depth Ca. 650 m), (Fahmy 2003).

The Gulf of Aqaba is a narrow deep trench extending from Lat. 28° N, Long 24° 23' E,

to Lat 29° 33' N, Long 35° 0' E'. It is about 180 Km long and has an average width of 20 Km and mean depth of about 800 m. it is bordered on both sides with a series of high mountains particularly along its western side.

Due to the importance of the Aqaba Gulf for tourism and the consequent important national income, the Environmental Information and Monitoring Program (EIMP) promoted the initiation of a consistent monitoring aims to the construction of a database system for the water quality, as the basis for the proper management of its coastal region. The EIMP was coordinated by a steering committee with representatives from International the Danish Development Assistance (Danida) and Egyptian Environmental Affair Agency (EEAA). The present study is a part of this program and aims at investigating the spatial and temporal pattern of the water quality along the Egyptian coasts of Agaba Gulf during the period from 1998 to 2004.

The strategic approach of the EIMP has been taken to initiate the monitoring programme and data base system for the coastal water quality of Aqaba Gulf by using the quality control and quality assurance works in order to evaluate the human impacts by time due to the rapid developments of the recreational and human activities on its coast.

## 2. MATERIALS AND METHODS

Within the frame work of EIMP, six field surveys have been performed annually from 1998 to 2004. Eleven stations were selected to represent all different environmental conditions in terms of human activities, public resort beaches and some reference sites (Fig. 1). Water temperature, salinity, dissolved oxygen (DO) and pH were measured in situ at each station using a CTD (YSI-600XL). Duplicate water samples for water quality variables were collected at 2-m depth (below surface water), using a PVC Niskin bottle. Ammonium ion concentrations were determined according to IOC (1983). Nitrite, Nitrate, reactive phosphate and reactive silicate concentrations were determined on pre-filtered seawater samples (Whatman GF/C) following the sepectrophotometric techniques described by IOC (1993) and Strickland and Parsons (1972). Total P and total N were estimated in unfiltered water samples following the procedure described by Valderrama (1981). The concentration of Dissolved Inorganic Nitrogen (DIN as the sum of  $NH_4$ -N +  $NO_2$ -N + NO<sub>3</sub>-N) was calculated. Total suspended matter (TSM) was collected from 3 L seawater samples by filtration through washed, dried and pre-weighed 0.45 µm membrane filter. The filters with the retained particles were washed then air dried in the oven at 60 °C for 24 – 48 hours until constant weight. The difference between the dry weight of membrane filters before and after filtration was expressed in mg/l. For chlorophyll-a (Chl-a) determination. additional 31 water samples were collected and filtrated on 0.45 µm filters. Chl-a was extracted using 90% acetone and measured spectrophotometerically according to Strickland and Parsons (1972). Water transparency was measured using Secchi Disk.

The correlation coefficients between different environmental parameters were calculated (n = 77 p  $\leq$  0.10) the correlation coefficient is significant at r  $\geq$  0.183.

AHMED M. ABDEL-HALIM et al



#### **3. RESULTS AND DISCUSSION**

#### 3.1. Hydrographical conditions

The ranges and mean values of water temperature, salinity, pH, dissolved oxygen (expressed in mg/l and % saturation) chlorophyll-a and total suspended matter reported in the present study during the period from 1998 to 2004 are shown in Table (1) and represented graphically in Fig. 2. The distribution of water temperature was accompanied with their geographic variations i.e. they followed changes in air temperature at different regions of the present study. Stations Aq1, Aq2 and Aq3 represented the highest values of water temperature during the period of study. The annual averages of water temperature varied from 22.65 to 26.55 °C (at station Aq11 and Aq2 during 2000 and 2001, respectively). A slight increase in water temperature was observed moving southward (from station Aq11 to Aq1). The lowest water temperature values were recorded during 2000, while the highest measurements were observed during 2001. The spatial distribution of salinity displayed a slight increase moving from south to north during the period of study except for 2002 and 2004. The highest mean value of salinity was recorded during 1998 (40.44) at station Aq9, while the lowest one at station Aq5 during 2002 (39.28) due to the relative increase of human impact in Dahab (Aq5). A significant negative correlation between salinity and water temperature was deduced  $(r = -0.32, n = 77, p \le 0.05)$ . pH and DO did not display clear changes during the period of study. The annual means of pH values were varied from 8.07 at station Aq6 during 2001 and 8.33 at Aq1 during 1999. The revealing distribution pattern of DO indicates high values and well oxygenated waters during the period of study (1998 - 2004). The annual means of DO ranged between 8.40 mg/l (equivalent to 128.15 % DO saturation) calculated at station Aq10 during 2004 and 6.82 mg/l (equivalent to 101.50 % saturation) calculated at station Aq5 during 2002. Minor changes of studied variables (water temperature, salinity, pH and DO) reveal that the effect of human impact on the distribution pattern of different hydrographical conditions in the Gulf of Agaba coastal water are still limited. This was expected due to the low population there, the absence of fresh water sources and limitation of land based source (i.e. sewage, agriculture, and / or industrial effluents). Accordingly, these conditions could be principally controlled by the circulation pattern of seawater in the area of There are significant positive study. correlation between salinity and each of pH and DO (r = 0.40, r = 0.37, n = 77,  $p \le 0.05$ respectively) and a moderated negative correlation between water temperature and DO (r = -0.15, n = 77, p  $\leq$  0.05). These conditions confirm the limited effect of human impacts on this region of the Red Sea.

#### **3.2.** Variables of Trophic State

# 3.2.1. Chlorophyll-a, Total Suspended Matter and Secchi Disk depth

The ranges and mean values of Chl-a, TSM are listed in Tables (1) and represented graphically in Fig. 2. Low Chl-a and TSM concentrations and high transparency were generally encountered at most stations. This can be highlighted by the distribution of mean values, at which, Chl-a ranged from 0.05 µg/l at station Aq10 during 2001 and 0.35 µg/l at station Aq5 during 1998 due to the relative increase of human impact in Dahab (Aq5). Total suspended matter ranged from 3.88 mg/l at station Aq3 during 1999 and 7.71 mg/l at station Aq7 during 2004. Finally, Secchi disk transparency ranged from 5.70 m at Aq2 during 2004 and 23.4 m at station Aq10 during 1998. Transparency reached bottom depth at most stations. TSM Chl-a showed a homogeneous and distribution at several locations. Their values

AHMED M. ABDEL-HALIM et al
----------------------------

were significantly correlated (r = 0.58, n = 77,  $p \le 0.05$ ) TSM measurements were negatively correlated with transparency (r = -0.24). Low levels of Chl-a and TSM coupled

with high values of secchi disk transparency clearly indicate that the effect of human impact in the investigated Aqaba Gulf coastal waters is still insignificant.

St. No.	water °C	Salinity	Hq	DO (mg/l)	Chl-a $(\mu g/l)$	TSM (mg/l)	Trans. (m)
Aq 1	24.48 - 26.35	39.34 - 40.04	8.15 - 8.33	6.83 - 7.86	0.09 - 0.25	5.08 - 7.01	9.30 - 12.00
	$(25.72 \pm 0.72)$	$(39.64 \pm 0.28)$	$(8.21 \pm 0.06)$	$(7.18 \pm 0.35)$	$(0.16 \pm 0.07)$	$(5.91 \pm 0.78)$	$(10.43 \pm 0.86)$
Aq 2	24.99 - 26.53	39.34 - 39.96	8.15 - 8.32	6.83 - 7.31	0.09 - 0.28	5.05 - 6.75	5.70 - 7.88
	$(26.11 \pm 0.65)$	$(39.65 \pm 0.24)$	$(8.21 \pm 0.06)$	$(7.06 \pm 0.20)$	$(0.17 \pm 0.07)$	$(5.94 \pm 0.53)$	$(6.54 \pm 0.73)$
Aq 3	24.82 - 26.33	39.30 - 40.02	8.15 - 8.32	6.89 - 7.48	0.10 - 0.30	3.88 - 6.56	9.24 - 10.34
	$(25.78 \pm 0.59)$	$(39.64 \pm 0.28)$	$(8.21 \pm 0.06)$	$(7.11 \pm 0.19)$	$(0.15 \pm 0.07)$	$(5.12 \pm 1.00)$	$(9.77 \pm 0.37)$
Aq 4	23.40 - 26.30	39.40 - 40.16	8.15 - 8.31	7.13 - 7.73	0.07 - 0.32	4.22 - 7.27	6.78 - 15.75
	$(24.99 \pm 0.85)$	$(39.75 \pm 0.29)$	$(8.20 \pm 0.06)$	$(7.37 \pm 0.23)$	$(0.16 \pm 0.09)$	$(5.61 \pm 1.09)$	$(10.52 \pm 2.73)$
Aq 5	23.28 - 25.13	39.28 - 40.13	8.13 - 8.30	6.82 - 7.70	0.07 - 0.35	4.42 - 6.93	7.42 - 19.70
	$(24.62 \pm 0.63)$	$(39.75 \pm 0.31)$	$(8.19 \pm 0.06)$	$(7.25 \pm 0.32)$	$(0.15 \pm 0.10)$	$(5.60 \pm 0.96)$	$(11.57 \pm 4.00)$
Aq 6	22.80 - 25.28	39.32 - 40.36	8.07 - 8.31	7.07 - 8.04	0.07 - 0.22	5.18 - 7.55	7.80 - 21.80
	$(24.50 \pm 0.83)$	$(39.83 \pm 0.35)$	$(8.20 \pm 0.07)$	$(7.37 \pm 0.32)$	$(0.13 \pm 0.05)$	$(6.05 \pm 0.91)$	$(15.56 \pm 5.88)$
Aq 7	23.13 - 25.64	39.36 - 40.36	8.09 - 8.30	7.06 - 7.96	0.06 - 0.20	4.70 - 7.71	8.50 - 16.60
	$(24.72 \pm 0.80)$	$(39.84 \pm 0.31)$	$(8.18 \pm 0.08)$	$(7.31 \pm 0.32)$	$(0.13 \pm 0.05)$	$(5.74 \pm 0.96)$	$(12.70 \pm 3.18)$
Aq 8	23.18 - 25.59	39.30 - 40.25	8.08 - 8.27	7.04 - 7.85	0.07 - 0.30	4.67 - 7.18	7.54 - 22.60
0	$(24.75 \pm 0.75)$	$(39.84 \pm 0.33)$	$(8.20 \pm 0.07)$	$(7.29 \pm 0.27)$	$(0.15 \pm 0.08)$	$(5.48 \pm 0.88)$	$(15.57 \pm 5.91)$
Aq 9	23.44 - 25.82	39.28 - 40.44	8.14 - 8.27	6.98 - 7.40	0.09 - 0.31	4.72 - 7.08	7.10 - 11.10
	$(25.03 \pm 0.79)$	$(39.87 \pm 0.36)$	$(8.19 \pm 0.05)$	$(7.12 \pm 0.15)$	$(0.16 \pm 0.08)$	$(5.78 \pm 0.90)$	$(8.94 \pm 1.55)$
Aq 10	22.85 - 25.44	39.32 - 40.20	8.16 - 8.29	6.82 - 8.40	0.05 - 0.17	4.38 - 7.37	8.72 - 23.40
	$(24.40 \pm 0.84)$	$(39.82 \pm 0.28)$	$(8.21 \pm 0.05)$	$(7.33 \pm 0.54)$	$(0.12 \pm 0.04)$	$(5.50 \pm 0.97)$	$(15.79 \pm 5.59)$
Aq 11	22.65 - 25.59	39.30 - 40.16	8.13 - 8.27	6.86 - 7.66	0.08 - 0.21	4.29 - 7.32	10.00 - 21.50
	$(24.36 \pm 0.95)$	$(39.81 \pm 0.29)$	$(8.18 \pm 0.06)$	$(7.27 \pm 0.30)$	$(0.15 \pm 0.05)$	$(5.54 \pm 0.96)$	$(17.19 \pm 4.21)$
Average	$(25.00 \pm 0.95)$	$(39.77 \pm 0.29)$	$(8.20 \pm 0.06)$	$(7.24 \pm 0.30)$	$(0.15 \pm 0.07)$	$(5.66 \pm 0.89)$	$(12.24 \pm 4.79)$

)04.
20
1
860
16
30
·E
qı
er
val
~
sta
03
õ
ace
II
SU
ulf
5
ba
gal
A
of
2
te
rac
lal
0
cal
"III
lei
SCI
IC.
iys
h
ed
ip
stu
f
S
Ine
val
()
IOI
at
S
p
ard
p
tar
s t
E
ea
Е
p
an
es
ng
ra
he
H
÷
-
ble
Ta
<u> </u>



Fig. (2): Variations in the annual mean values of some studied physicochemical characteristics of Aqaba Gulf Coastal Water during 1998 – 2004.

#### 3.2.2. Nutrient salts

The hydro-chemical characteristics of the Red Sea water depend on its dynamics as well as on the geographical location. The Red Sea does not receive any significant nutrient supply from river out flow.

The mean values of different nutrients Aqaba Gulf coastal water are listed in Table (2) and represented graphically in Figs. 3, 4. They indicated that, dissolved inorganic nitrogen concentrations during the seven years of study are quite low. This is evidence from the mean values of NH<sub>4</sub>-N, which varied from 0.06 µM at station Aq9 during 2001 to 1.78 µM at station Aq2 during 2000. Nitrite concentrations were always less than 0.1 µM in all stations during the study period except some stations of year 2004 which exceeds this level. The maximum annual average (0.19 µM) was recorded at station Aq11 during 2004. Nitrate concentrations ranged from 0.08 to 2.19 measured during 1999 at station Aq2 and Aq11, respectively. Based on the mean values, the concentrations of dissolved inorganic nitrogen forms followed the order  $NH_4 > NO_3 \ge NO_2$  as shown in the following table:

Year	NH <sub>4</sub> %	NO <sub>3</sub> %	NO <sub>2</sub> %
1998	63.46	33.80	2.74
1999	69.62	28.24	2.14
2000	45.42	51.07	3.15
2001	48.27	49.17	2.53
2002	68.48	27.78	3.74
2003	59.78	34.88	5.33
2004	46.15	42.00	11.85

The relative increase of  $NH_4$  concentration which is the preferred N form for phytoplankton uptake is unclear. A

significant positive correlation between NH<sub>4</sub>-N and Chl-a was found during the study period (r = 0.23). These concentrations, based on the classification reported by Skrivanic and Strin (1982) and Franco (1983), allow classifying the Aqaba Gulf coastal waters as oligotrophic to mesotrophic. These authors indicated as oligtrophic, seawater displaying concentration of 0.5  $\mu$ M for each NH<sub>4</sub> and NO<sub>3</sub> whereas, in eutrophic water the concentration of these nutrients are usually in the order of 2.0  $\mu$ M for NH<sub>4</sub> and 4  $\mu$ M for NO<sub>3</sub>.

The levels of total nitrogen (TN) displayed remarkable variations (from 2000 to 2004) and ranged from 13.62 to 59.08  $\mu$ M measured at station Aq7 during 2000 and at station Aq11 during 2004, respectively. Based on the mean values, DIN accounted about 6 % of TN (8.23 -4.34%). The large difference between ΤN and DIN concentrations suggest that nitrogen is found in the Agaba Gulf coastal waters mostly in organic forms. This result is in agreement with the general view of microbial food web and phytoplankton dynamics, in which NH<sub>4</sub>, NO<sub>2</sub> and NO<sub>3</sub> are rapidly processed by and other phytoplankton microbial components. Meanwhile, the organic nitrogen is assimilating by aquatic organisms in a much slower rate (Riley and Chester 1974). Faganeli (1983) pointed out that, in the eutrophic Bay of Koper (North Adriatic), the relative composition of total nitrogen are 11.3 % for particulate, 68.8 % for dissolved organic and 20 % for the inorganic forms. DIN levels obtained during the present investigation are remarkably lower than those mentioned above coincided with the limited effect of land based source on the Agaba Gulf coastal waters. A nonsigneficant negative correlation between DIN and DO was deduced (r= -0.15, n = 77, p  $\leq$  0.05).

St. No.	NH4-N	NO2 -N	NO3 - N	IN	PO4-P	TP	Si04 - Si
Aq 1	0.61 - 1.68	0.02 - 0.13	0.18 - 0.87	12.86 - 22.77	0.01 -0.08	0.41 - 1.82	0.71 - 1.33
	$(0.93 \pm 0.39)$	$(0.05 \pm 0.04)$	$(0.49 \pm 0.25)$	$(18.99 \pm 4.26)$	$(0.04 \pm 0.02)$	$(0.91 \pm 0.50)$	$(0.98 \pm 0.22)$
Aq 2	0.64 - 1.78	0.02 - 0.12	0.08 - 0.60	17.63 - 35.06	0.02 - 0.05	0.54 - 2.24	0.72 - 1.28
	$(1.16 \pm 0.43)$	$(0.05 \pm 0.03)$	$(0.39 \pm 0.17)$	$(23.52 \pm 7.09)$	$(0.04 \pm 0.01)$	$(1.08 \pm 0.62)$	$(1.16 \pm 0.20)$
Aq 3	0.47 - 1.39	0.02 - 0.13	0.19 - 0.76	15.67 - 32.97	0.02 - 0.08	0.44 - 2.19	0.75 - 1.42
	$(0.79 \pm 0.32)$	$(0.05 \pm 0.04)$	$(0.41 \pm 0.20)$	$(24.25 \pm 6.60)$	$(0.04 \pm 0.02)$	$(1.10 \pm 0.60)$	$(1.16 \pm 0.25)$
Aq 4	0.53 - 1.40	0.02 - 0.12	0.34 - 0.69	19.47 - 26.12	0.01 - 0.07	0.56 - 1.84	0.57 - 1.38
	$(0.84 \pm 0.37)$	$(0.04 \pm 0.03)$	$(0.51 \pm 0.14)$	$(23.37 \pm 2.53)$	$(0.03 \pm 0.02)$	$(0.92 \pm 0.44)$	$(1.10 \pm 0.27)$
Aq 5	0.44 - 1.28	0.02 - 0.15	0.23 - 0.78	19.12 - 37.76	ND - 0.12	0.55 - 1.72	0.54 - 1.88
	$(0.75 \pm 0.28)$	$(0.05 \pm 0.05)$	$(0.46 \pm 0.17)$	$(28.33 \pm 8.05)$	$(0.04 \pm 0.04)$	$(1.02 \pm 0.44)$	$(1.08 \pm 0.42)$
9 bV	0.39 - 1.23	ND - 0.13	0.25 - 1.11	15.71 - 34.79	0.02 - 0.27	0.51 - 2.21	0.75 - 1.56
	$(0.66 \pm 0.28)$	$(0.05 \pm 0.04)$	$(0.60 \pm 0.30)$	$(25.65 \pm 6.89)$	$(0.07 \pm 0.09)$	$(1.21 \pm 0.52)$	$(1.11 \pm 0.26)$
Aq 7	0.43 - 1.31	0.01 - 0.13	0.36 - 1.15	13.62 - 32.77	0.01 - 0.13	0.53 - 2.45	0.56 - 1.38
	$(0.77 \pm 0.33)$	$(0.06 \pm 0.04)$	$(0.53 \pm 0.29)$	$(25.52 \pm 7.31)$	$(0.06 \pm 0.01)$	$(1.14 \pm 0.68)$	$(1.05 \pm 0.27)$
Aq 8	0.33 - 1.39	0.01 - 0.13	0.16 - 0.55	14.34 - 39.47	0.01 - 0.42	0.61 - 4.85	0.77 - 1.61
	$(0.73 \pm 0.40)$	$(0.06 \pm 0.04)$	$(0.39 \pm 0.14)$	$(26.18 \pm 9.25)$	$(0.1 \pm 0.15)$	$(1.62 \pm 1.49)$	$(1.07 \pm 0.29)$
6 bY	0.06 - 0.98	0.01 - 0.13	0.15 -0.81	21.64 - 49.71	0.03 - 0.08	0.45 - 1.85	0.57 - 1.53
	$(0.59 \pm 0.33)$	$(0.05 \pm 0.04)$	$(0.48 \pm 0.21)$	$(28.46 \pm 11.96)$	$(0.04 \pm 0.02)$	$(0.98 \pm 0.44)$	$(0.93 \pm 0.32)$
Aq 10	0.37 - 0.96	0.04 - 0.15	0.18 - 0.57	15.68 - 28.31	0.02 - 0.07	0.49 - 2.37	0.49 - 1.14
	$(0.67 \pm 0.22)$	$(0.08 \pm 0.04)$	$(0.42 \pm 0.12)$	$(22.84 \pm 4.77)$	$(0.04 \pm 0.02)$	$(1.00 \pm 0.65)$	$(0.87 \pm 0.26)$
Aq 11	0.32 - 1.12	0.05 - 0.19	0.40 - 2.19	18.33 - 59.08	0.02 - 0.23	0.64 - 2.51	0.60 - 1.40
	$(0.73 \pm 0.28)$	$(0.09 \pm 0.05)$	$(1.10 \pm 0.66)$	$(32.65 \pm 15.65)$	$(0.07 \pm 0.07)$	$(1.32 \pm 0.66)$	$(0.98 \pm 0.29)$
Average	$(0.78 \pm 0.35)$	$0.06 \pm 0.04)$	$(0.53 \pm 0.33)$	$(25.43 \pm 8.34)$	$(0.05 \pm 0.06)$	$(1.12 \pm 0.68)$	$(1.04 \pm 0.28)$

Table (2): The ranges and (mean  $\pm$  standard deviation) values of the nutrient salts of Aqaba Gulf surface coastal water during 1998 – 2004.



Fig. (3): Variations in the annual mean values of nitrogen forms of Aqaba Gulf Coastal Water during 1998 – 2004.



Fig. (4): Variations in the annual mean values of phosphate and reactive silicate of Aqaba Gulf Coastal Water during 1998 – 2004.

A remarkable increase in the mean values of phosphate concentrations was observed at stations Aq8 (0.19 µM) during 1998, Aq11 (0.23 µM) during 1999 and Aq7, Aq11 (0.13 and 0.09 µM, respectively) during 2000 and Aq5, Aq6, Aq7 and Aq8 (0.12, 0.27, 0.11 and 0.42 µM) during 2002. These stations are located in the area of Newabea port and Taba City which affected directly with the relative increase in human activities. At the other locations, PO<sub>4</sub> were very low, consequently the N:P ratios were very high reaching 78 : 1 and suggesting that, phosphate could be the limiting nutrient for phytoplankton growth. A strong positive correlation between phosphate and DIN was calculated (r = 0.41, n = 77, p  $\leq$ 0.05). The variation in PO<sub>4</sub> concentrations was associated with the level of impact on the Newabea Port area of Aqaba Gulf coastal waters. Low PO<sub>4</sub> contents could be related mostly to their sorption and deposition on iron born dust conveyed to the basin from the surrounding mountains and deserts. Suzumura et al (2000) reported the effect of composition and physicochemical characteristics of natural particles on phosphate adsorption-desorption processes under various aquatic environment. Marchetti (1984) pointed out that, generally the concentration of PO<sub>4</sub> in the surface waters are extremely low (expressed as values for orthophosphate 0.03 µM or less), whereas typical concentrations for eutrophic coastal waters are above 0.15 µM and highly eutrophic system will be beyond 0.3 µM. Okbah et al 1999 reported that inorganic phosphorus concentrations in Aqaba Gulf are very low particularly in the euphotic zone (~ 200 m). In contrast, the Northern Red Sea has higher values ranged between not detected - 0.34 temporal The geographic and μM. distribution pattern of TP displayed a large variability during the investigation period TP ranged from 0.41 to 4.85 µM at station Aq1 and Aq8 during 2001 and 1998, respectively. The highest TP values were measured during 1998 (Fig. 4). Based on the mean values, phosphate constituted about 5.1 % of TP, implying that phosphate is mostly accounted by particulate and organic forms. These data also suggested that the coastal water of Agaba Gulf is unpolluted. Such conclusion is consistent with indications provided by Nalewajko and Lean (1980) who pointed out that in moderated polluted coastal waters, the relative composition of phosphorus forms are: 28.5 98% for particulate, 1.2 - 4% for colloidal, 0.1 - 22 % for reactive phosphate and 0.1 - 6 % for dissolved organic P. Giovanardi and Tromellini (1992) stated that the levels of TP and TN in oligotrophic waters are 0.27 and 47.2 µM, whereas in mesotrophic waters reach 0.89 and 53.8 µM and in eutrophic seawaters 2.81 and 133.9 µM. these levels when compared with those reported in the present study indicate that the Aqaba Gulf coastal waters in front of Egypt are "located" within the oligo to meso-trophic conditions.

The distribution pattern of SiO₄ concentration display small spatial annual variation. The mean values ranged from 0.49 to 1.88 at Aq10 during 2003 and Aq5 during 1999, respectively. (Fig. 4). Generally, the highest values determined during 1999, the lowest values during 2002. Okbah et al (1999) pointed out those surface values of silicate in the Aqaba Gulf are lower than those of the Northern Red Sea water. Beltagi (1984) pointed out that, primary producers of northern Red Sea are mainly composed of blue green algae to a lesser extent by diatoms, the main consumer of silicate. This means that, diatoms constitute a minor component in the Aqaba Gulf and Northern Red Sea. According to Fahmy (2003) Stated that, the main factors controlling SiO<sub>4</sub> distribution in the Egyptian Red Sea coastal waters, are:

i) the supply of  $SiO_4$  which flows in the Red Sea through the straight of Bab El-Mandab

ii) biological composition iii) organic matter decomposition and

iv) the partial dissolution of quartz and clay particles transported to the sea from the surrounding deserts during sand storms. Silicate was correlated positively with each of water temperature, salinity, pH and DO (r = 0.19, 0.26, 0.52 and 0.23, respectively, n = 77, p  $\leq$  0.05), and negatively correlated with each of ammonia, nitrite, DIN and phosphate (-0.19, -0.19, -0.19 and -0.17 respectively).

Significant positive correlations were signified between Chl-a and each of salinity, DO, TSM, ammonia, nitrite, DIN and TP (r = 0.37, 0.28, 0.58, 0.23, 0.27, 0.22 and 0.35 respectively,  $n = 77, p \le 0.05$ ).

#### **4. CONCLUSION**

The data obtained during the present study signified that, monitoring and data base system constructed for Aqaba Gulf Coastal waters in front of Egypt during seven years (1998 - 2004) revealed no significant variations for each of salinity and pH values, well oxygenated seawater, low levels for each of Chl-a, TSM, nitrogen, phosphorus and reactive silicate, accordingly, it can be classified between oligotrophic to mesotrophic state. Therefore, it is safe to conclude that the main body of Aqaba Gulf coastal regions in front of Egypt is not yet seriously threatened in spite of the rapid recreational and human developments taken place on its coast during the previous ten vears.

#### ACKNOWLEDGEMENTS

The support of Danish International Assistance (Danida) and Egyptian Environmental Affair Agency (EEAA) are acknowledged.

## REFERENCES

- Anati, D.A.: 1974, Water transports in the Gulf of Aqaba, L'Oceanographie Physique de la Mer Rouge. CNEXO Publ. No 2 Ser. Act. Colloq., IAPSO Symp., Paris, pp 165 – 173
- Beltagi, A. I.: 1984, Oceanographic and Fisheries Investigations in the Egyptian Red Sea. Special publication, Academy of

Scientific Research and Technology, NIOF Egypt, p 98.

- Faganeli, J.: 1983, Organic nitrogen and phosphorus in Gulf of Trieste (N. Adriatic) Archives Oceanography and Limnology, 20: 153 – 117.
- Fahmy, M. A.: 2003, Water quality in the Red Sea coastal waters (Egypt): Analysis of spatial and temporal variability. *Chemistry and Ecology*, Vol. **19** (1), pp 67 – 77.
- Franco, P.: 1983, Fattori influenti sulla produttivita primaria dell'Adriatico settentrionale. Proceeding International Conference: Problems of Adriatic Sea Trieste, Vol. **155**, p A4.
- Giovanardi and Tromellini, E. (1992) Statistical assessment of tropic conditions. Marine Eutrophication Proceedings of an International Conference Bologna, Italy, 21 – 24 March 1990, pp 211 – 233.
- International Oceanographic Commission: 1983, Chemical methods for use in marine environmental monitoring. Manuals and Guides. UNESCO, p 53.
- International Oceanographic Commission: 1993, Nutrient analysis in tropic marine waters. Manuals and Guides. Vol. **28** UNESCO, pp. 1 24.
- Marchetti, R.: 1984, Quadro analitico complessivo del risultati delle indagini condotte negli anni 1977 – 1980. II. Problema dell'eutrofizzazione nelle acque costiere dell'Emilia- Romagna: Situazione ipotesi di intervento, Bologna, Regione Emilia Romagna, p 308.
- Nalewajko, C. and Lean, D. R.: 1980, Phosphorus. In: Morris(Ed), The physiological Ecology of Phytoplankton, Blackwell Science Publisher, Oxford, pp 235 – 258.
- Okbah, M. A.; Mahmoud, Th. H. and El-Deek, M. S.: 1999, Nutrient salts concentrations in the Gulf of Aqaba and Northern Red Sea. *Bull. Nat. Inst. Of Oceanogr.* & Fish. Vol. 25: 103 – 116.
- Reiss, Z. and Hottinger, L.: 1984, The Gulf of Aqaba. Ecological Micropaleontology.

Spring- Verlag, Berlin, Heidelberg, New York, Tokyo, p 330

- Riley, J. P. and Chester, R.: 1971, Introduction to marine chemistry, Academic Press, p 757.
- Sharaf El-Din, S. H. and Soliman, G. F.: 1995, Water movement variabilities in the Gulf of Suez and Gulf of Aqaba as two legs of the Red Sea. First International Conference on the Present and Future Technology of Navigation and Marine Science of Mediterranean and Red Sea. 29 -31 October 1995, El-Manharousa, Alexandria, Egypt.
- Skrivanic, V. A. and Strin, J.: 1982, Basic Physical Chemical and Biological Data Reports R. V. A Mohorov ICIC Adriatic Cruises 1974 – 76. Hydrographic Institute of Yugoslav Navy split, p. 175.
- Soliman, G.F.: 1995, Water movements in the Gulf of Aqaba as homogeneous basin, derived by tidal and wind forces. First International Conference on the Present and Future Technology of Navigation and Marine Science of Mediterranean and Red Sea. 29 -31 October 1995, El-Manharousa, Alexandria, Egypt.

- Strickland, J. D. H. and Parsons, T. R.: 1972, A Practical Hand Book of Sea Water Analysis. Fisheries Research Board of Canada Bulletin 167, 2<sup>nd</sup> ed., p. 310.
- Suzumura, M., Ueda, S. and Suni, E.: 2000, Control of phosphate concentration through adsorption and desorption process in ground water and sea water mixing at sand beaches in Tokyo Bay. *Japan Journal of Oceanographic Society of Japan*, **56**. pp 667 – 673.
- Valderrama, J. C.: 1981, The simultaneous analysis of total nitrogen and total phosphorus in natural waters. *Marine Chemistry*, **10** pp 109 – 122.