

RELATION BETWEEN SEDIMENT CALCIUM CARBONATE, ORGANIC MATTER AND PRODUCTIVITY IN ASWAN HIGH DAM RESERVOIR.

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ABSTRACT

The purpose of the survey of the Newly-formed Aswan High Dam Reservoir is to indicate a basis for a determination of calcium carbonate and organic matter in the bottom sediments and its effects on the reservoir productivity. This has been executed by correlating the calcium carbonate, organic matter, plankton content, fish population and bottom fauna at three selected sectors (northern, middle and southern) that almost represent the whole reservoir. The results have shown enrichment in productivity in the middle area of the lake. There was a significant correlation between calcium carbonate concentration in the bottom sediments with the reservoir productivity.

INTRODUCTION

The present work deals with the organic matter and calcium carbonate in the sediments of the Aswan High Dam Reservoir, chlorophyll "a" and the bottom fauna. Thirty sediment samples representing the area under investigation in the three cross sections, viz. El-Birba, 10 Km; Amada 200 Km and Adingan 300 Km, south of the High Dam, respectively.

MATERIAL AND METHODS

The present investigation covered the period from March 1981 to March 1983. During this time the studies included some chemistry of the sediments, bottom fauna, and chlorophyll "a". These studies were made at three sections representing the various parts of the reservoir (Fig. 1). Sampling was carried out at seasonal intervals; the Ekman dredge, was used for sediment collection. The organic matter was determined by oxidation with chromic acid as described by Hanna (1965). The calcium carbonate was determined, using the method given by Jackson (1958). Chlorophyll "a" was determined according to Lind (1974).

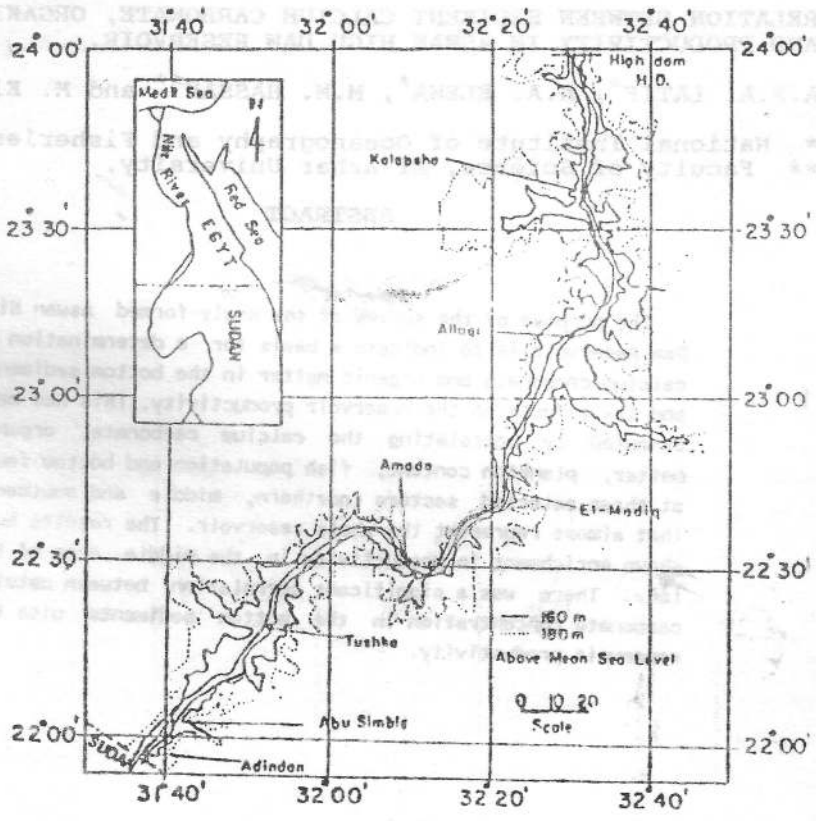


FIG. 1
Location of stations in Aswan High Dam Reservoir

RESULTS

Chlorophyll "a" :

As given in Table 1 and Fig. 2 B, Amada water shows the highest values of chlorophyll "a" (11.1 ug/l for the winter and 11.7 ug/l for spring) as compared with 6.1-6.7 ug/l in the periods (autumn and the summer). Again the lowest value is recorded in summer at El-Birba.

Zooplankton:

The zooplankton concentration in Lake Nasser increased southwards. At the same time its highest frequency was recorded in the spring, followed by the Autumn (Table 2). The least frequency was confined to reworkers in the summer and autumn at El-Birba and in the winter at both Amada and El-Birba (Fig. 3).

Table 1
 Chlorophyll "a" concentration (ug/l) in Aswan
 High Dam Reservoir during 1981-1983

Location	Seasons			
	Spring	Summer	Autumn	Winter
El Birba	3.8	0.95	4.7	5.5
Amada	11.7	6.1	6.7	11.1
Adindan	2.8	3.5	3.3	2.0

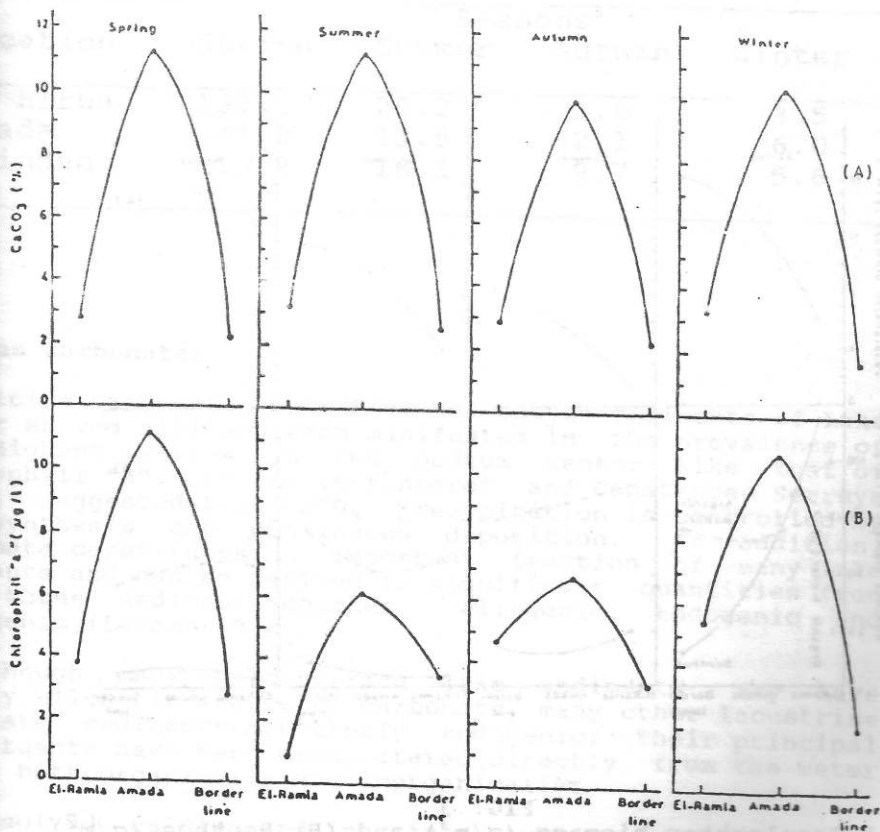


FIG. 2
 Calcium carbonate in the bottom sediment (A)
 and chlorophyll "a" (B) in Lake Nasser water

Table 2
 Zooplankton biomass (g/m^3) in Aswan
 High Dam Reservoir during 1981-1983

Location	Seasons			
	Spring	Summer	Autumn	Winter
El Birba	6.1	1.5	4.1	3.2
Amada	13.3	5.9	9.7	4.8
Adindan	16.4	8.3	11.8	5.5

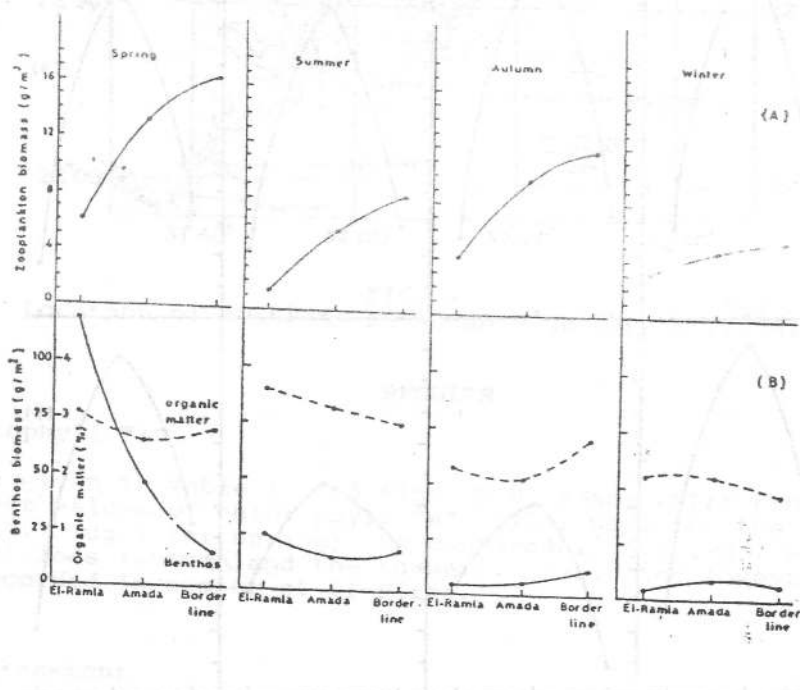


FIG. 3
 (A) Zooplankton biomass (g/m^3) and (B) Benthos (g/m^2)
 and organic matter (%) in the sediments of Lake Nasser

Benthos:

The benthic animals in the sediments of Lake Nasser include mainly oligochaete, worms, larvae of insects and bivalve molluscs. As shown in Table 3 and Fig. 3, the concentration of the benthos in the spring was much more abundant at El-Birba than either at Amada or at Adindan. In autumn, the highest benthos biomass prevailed at Adindan followed by Amada, while the least value was recorded at El-Birba.

Table 3
Benthos biomass (g/m^2) in Aswan
High Dam Reservoir Sediment (1981-1983)

Location	Seasons			
	Spring	Summer	Autumn	Winter
El Birba	139.5	25.2	1.0	3.5
Amada	44.0	13.8	2.1	6.0
Adindan	13.2	16.1	9.7	5.6

Calcium carbonate:

Calcium carbonate content of the bottom sediments of Lake Nasser showed similar trend manifested in the prevalence of the highest values in the medium sector like that of chlorophyll "a". In Lakes Kinneret and Constance, Serruya (1971) suggested that $CaCO_3$ precipitation is controlled by photosynthesis and continuous deposition. In addition, carbonate constitutes an important fraction of many lake sediments and can be derived in significant quantities from the three sediment sources; alloctogenic, endogenic and lithogenic (Lerman 1978).

Although many calcite-rich lake sediments may have largely alloctogenic source of carbonate, many other lacustrine carbonate sediments are truly endogenic; their principal constituents have been precipitated directly from the water column both organically and inorganically.

Dissolved calcium and inorganic carbon can be supplied through dissolution of older carbonate rocks in the drainage area, or by hydrolysis/carbonation reactions with calcium bearing silicates.

Besides requiring a source of sufficient calcium and dissolved carbonate for precipitation, numerous factors within the Lake can affect the solubility and thus precipitation of carbonate. Both organic and inorganic mechanisms are believed responsible for carbonate precipitation. Inorganically, the effect of temperature and CO_2 on solubility of the calcite was noticed by Kindle (1929). Higher water temperature, characteristic of shallow water favors precipitation through decreased solubility of both CaCO_3 and CO_2 . In Lake Nasser, Elewa (1976) mentioned that no carbonate could be detected in the bottom water in the warm and hot months of April and July, while concentration of 11.6 and 8 mg/l were recorded in cold seasons viz. November and January.

In the Aswan High Dam Reservoir sediments, the highest values of CaCO_3 were recorded at Amada (9.99 - 11.27 %) compared with 1.71 - 2.61 % and 2.81 - 3.37 % at Adindan and El-Birba, respectively. Generally, the CaCO_3 content in the sediments was higher during winter and spring than in autumn and the summer (Table 4 & Fig. 2). Thus, according to Lerman (1978), the relative role of organic and inorganic processes in determining the nature of lacustrine carbonate has not been entirely sorted out. Plants and planktonic algae are well known for their ability to extract CO_2 , raise pH and thus promote precipitation of carbonate. Phytoplankton production can be a major factor in the precipitation of calcium carbonate. Among others, Wetzel (1966) referred to colloidal carbonate precipitation resulting from organic productivity.

According to Golterman (1975), carbonate from the sinking dead material is partially converted into CO_2 . Furthermore, when the oxygen is depleted, fermentation processes may occur yielding organic compounds. Finally, part of the carbon may even be reduced to methane. On the other hand, the CO_2 produced will lead to a lower pH value which may cause dissolution of CaCO_3 . In Lake Nasser, the free CO_2 in the bottom water fluctuated between 0.0 - 3.3, 0.95 - 2.4, 0.0 - 3.2 and 0.0 - 1.7 mg/l during April, July, November and January, respectively compared with the zero value in the surface layers. At the same time, the pH values of Lake Nasser varied for the bottom water from 7.96 - 8.16, 7.58 - 7.58, 7.58 - 7.85 and 7.73 - 8.16 for the above months, respectively compared with the surface water (8.33 - 9.25, 8.45 - 9.13, 7.85 - 8.98 and 7.85 - 8.42, respectively). In addition, in calcium rich waters calcium carbonate may precipitate as the pH rises during photosynthesis. The precipitate will sink until it reaches the layers at the bottom with lower pH values resulting from CO_2 release, where it will dissolve again. The process may take place near the mud and can not therefore be distinguished from direct dissolution of CaCO_3 from the mud. Finally, the mediam part is known by its high productivity (Feed, Personal communication). The lowest value of the northern part may have been resulted from the fact that it is the deepest part of the Lake and characterized by the presence of the least thick oxygenated layer in the summer.

Table 4
Calcium carbonate (%) in the sediment
of Lake Nasser

Location	Seasons			
	Spring	Summer	Autumn	Winter
El Birba	2.81	3.28	2.97	3.37
Amada	11.26	11.27	9.99	10.44
Adindan	2.22	2.61	2.23	1.71

Needless to mention that the maximum value of CaCO_3 in Lake Nasser is much lower than that of the shallower more productive natural lakes of Egypt, as Lake Maruit where CaCO_3 content ranges from 14.8 to 68.8 % (El-Wakeel and Wahby, 1970).

Organic Matter:

In the marine sediments, the organic carbon ranges between less than 0.1 % and 30 % and is used as an index of the amount of food available to benthic animals or as an indication of the amount and type of food setting to the sediments from the water column (Ballinger and McKee, 1971 and Byers et al., 1978). At the same time, Round (1958) showed that Ca, pH and organic matter of the sediments could be used as nutrient index. Decomposition of organic material exerts oxygen demand, releases nutrients and contributes with H_2S and methane to water (Sylvester & Anderson, 1964). Thus, decomposition of the organic matter and the attendant oxygen demand are the result of the microbiological activity. Zobell (1946), distinguished between reducing intensity (measured by the redox potential) and reducing capacity (the rate of O_2 uptake).

The activities of animals in the deposits are often overlooked. Many invertebrate groups are represented with insect larvae, Mollusca and Oligochaete worms quantitatively predominant. Like earthworms, their feeding activities result in a through "working" of the sediments, help to break down organic matter and account for the absence of stratification in many deposits (Alsterberg, 1922).

Only a few specialized species (e.g. some chironomid larvae and Tubifex) can adapt themselves to the low oxygen content prevalent at the mud surface in shallow lakes in summer.

The distribution of species, numbers and weights with water depth is closely related to plant cover, particle size, and organic content of the deposit and the oxygen conditions in the hypolimnion. As with plankton production, high productivity is generally only found in shallow lakes. These bottom organisms form a part of food of the fish.

In Lake Nasser sediments, the organic matter fluctuated between 1.96 % and 2.23 %, 2.68-3.11, 3.03-3.62 and 2.18-2.87 % in spring, summer, autumn and winter, respectively. Generally, the highest values of organic matter recorded in the northern section were followed by or decrease towards the south (Table 5).

In conclusion, only the calcium carbonate concentration in the bottom sediment of Lake Nasser can be indirectly helpful in prognosing the primary productivity of the lake.

Table 5
Organic matter (%) in the bottom sediments
of Lake Nasser during 1981-1983

Location	Seasons			
	Spring	Summer	Autumn	Winter
El Birba	2.23	3.11	3.62	2.36
Amada	2.22	2.68	3.39	2.18
Adindan	1.96	2.85	3.03	2.87

REFERENCES

- Alsterberg, G., 1922. Die respiratorischen Mechanismen der Tubificiden-Acta Univ. Lund. (Lunds Univ. Arsskr.), NF Avol. 2, 18, (1): 175-604.
- Ballinger, D.C., and G.D. McKee, 1971. Chemical characterization of bottom sediments. J. Water Poll. Contr. Fed., 4 (2): P. 216.
- Byers, S.C.; E.L. Mills, and P.L. Stewart, 1978. A comparison of methods of determining organic carbon in marine sediments, with suggestion for standard method. Hydrobiol., 58 (1): 34-47.
- El-Wakeel, S.K. and S.D. Wahby, 1970. Texture and chemistry of Lake Maryut Sediments. Arch. Hydrobiol., 67 (3): 368-395.
- Elewa, A.A., 1976. Chemical Analysis for Lake Nasser. M.Sc. Thesis, Fac. of Sci. Al-Azhar University.

Golterman, H.L., 1975. **Physiological Limnology**. Elsevier scientific publishing Co.

Hanna, A. 1965. Organic matter in soil. In. **chemistry of soil**. 2nd (ed) F.E Bear. American chemical society monograph series. New York.

Jackson, M.L., 1958. **Soil chemical analysis**. Constrable, London.

Kindle, E.M., 1929. A comparative study of different types of thermal stratification in Lakes and their influence on the formation of Marl. *J. Geol.*, 37: 150-157.

Lind, O.T., 1974. **Common methods in limnology**. Mosby.

Lerman, A., 1978. **Lakes, Chemistry, Geology, Physics**. Springer Verlag, New York, 356 pp.

Round, F.E., 1958. Algal aspects of lake typology. *Verh, Int. Ver. Limnol.* 13: 306-316.

Serruya, C., 1971. Problems of sedimentation in the Lake of Geneva. *Verh. Int. Verein. Limnol.*, 17: 208-217.

Sylvester, R.O. and G.C. Anderson, 1964. A lake's response to its environments. *Jout. San. Eng. Div., Proc. Amer. Soc. Civil Engr.*, 90.

Wetzel, R.G., 1966. Productivity and nutrient relationships in marl lakes of northern Indiana. *Verb. Int. verein Limnol.*, 16: 321-332.

Zobell, C.E., 1946. **Marine Microbiology**. Waltham, Mass., Chronica Botanica, XV, 240 - pp. 868.

INTRODUCTION

The physiology, behavior and reproduction of crustaceans are linked to the molting cycle. Therefore, in experimental studies it is necessary to be able to identify completely the stages of the molting cycle in order to interpret observed physiological and biological changes.

Earlier studies of the molting stages of crustaceans have been carried out on several species including *Diaptomus* (Schäfer, 1968) *Pennaeus indicus* (Wood, 1968) *Pennaeus californianus* and *Pennaeus stylirostris* (Wood and Calvin, 1979) and *Pennaeus marginatus* (Longwell, 1961). The molting behavior of *Pennaeus aciculatus* has been described by (Warrenberg and Nil, 1984).

In crustaceans, eyestalk factors participate in the regulation of the direction of these activities at various stages in the molting cycle (McMahon and Chao, 1961). On the other hand, there are quite a lot of work on the effect of eyestalk removal or injections of extracts from those organs upon main physiological and biochemical processes in crustacea (McMahon et al., 1962; Chao et al., 1972; Bagnicker and Madhupratna, 1971).