

CHEMICAL CONDITIONS IN BARDAWIL LAGOON IV - NUTRIENT SALTS

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ABSTRACT

Water samples were collected from the Bardawil lagoon during the period from August 1985 to November 1986 for studying the distribution of nutrients. Nitrite showed a depletion nearly the year-round. Silicate was present only during summer, Ammonia showed a slight decrease towards the bottom, while nitrate showed a reverse pattern. The low N:P ratio (3:1) or (6 & 4 :1) indicates a faster assimilation of nitrate than of phosphate. The high ratio Si: P (35 : 1) and (23 :1) for both surface and bottom waters respectively indicates a stacking of diatom and regeneration of silicate.

INTRODUCTION

The ecosystem of Sabkhet El-Bardawil has been described elsewhere (Yitzhak, 1971; Siliem, 1989 a & b). Mediterranean Sea water inflows through Bughase I at the western arm. Bughase II was dugout during the period of this study at the eastern arm (Fig. 1). No streams flow into the Lagoon.

MATERIAL AND METHODS

Twelve stations covering the whole Lagoon were selected (Fig. 1). Water samples were collected at the surface and at the layer overlying the Lagoon's bottom by means of a standard water sampler (Ruttner sampler). Samples were analysed for ammonia, nitrite, nitrate, phosphate and silicate. These nutrients were determined according to Anon. (1965).

RESULTS AND DISCUSSION

The concentration of ammonia in the investigated area showed considerable variations and rapid changes (Fig. 2). The seasonal average values of ammonia in the Lagoon water ranged from 19.08 - 41.48 $\mu\text{g/l}$ at surface to 16.64 - 37.76 $\mu\text{g/l}$ at the subsurface layer. While the annual variations were 24.71 $\mu\text{g/l}$ and 22.5 $\mu\text{g/l}$ at the surface and near the bottom respectively. The ammonia content decreased downward. In addition, ammonia increased during summer (high temperature) and decreased in the other seasons (low temperature).

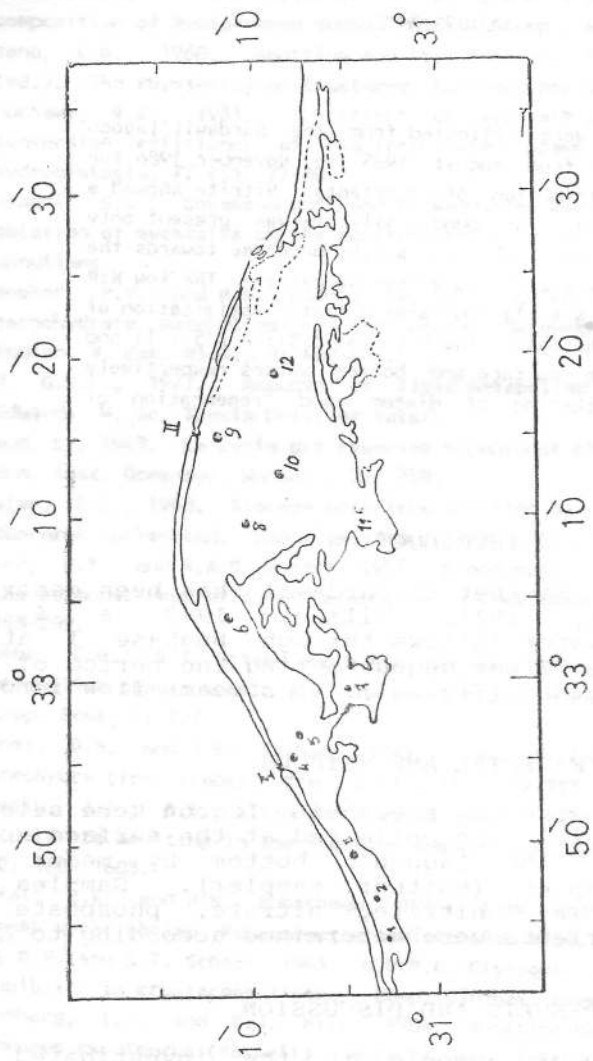


FIG. 1
Location of stations in Bardawil Lagoon

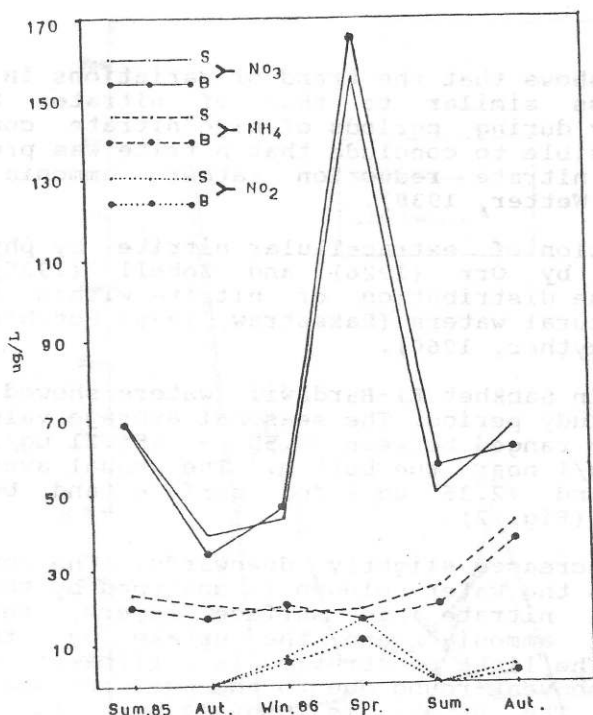


FIG. 2
Ammonia, Nitrite and Nitrate variations
during the years, 1985 and 1986

Sverdrup et al. (1942) found that ammonia in sea water shows a wide range in value from 4.9 to 49.03 ug/l. The values obtained for the Bardawil Lagoon were slightly lower than these values and more or less than the values given by Grasshoff (1976). He, stated that, the amount of NH₄ and NO₂ rarely exceed 70 ug/l in oxygenated, unpolluted waters and the toxic one to the fish is NH₃ and not NH₄ ion.

Nitrite in the Bardawil Lagoon showed a complete depletion during the period of investigation. Some exceptions were recorded; in winter at stations 1-5, spring at stations 2-4 and 6-7 for surface layer only and during autumn. Grasshoff (1976) confirmed that nitrite may be excreted by phytoplankton especially during periods of luxury feeding, i.e. if a surplus of nitrate and phosphate stimulates a heavy bloom of plankton (Fig. 2).

Figure 2 shows that the trend of variations in nitrite is more or less similar to that of nitrate. Nitrite was detected only during periods of high nitrate content which makes it possible to conclude that nitrite was produced as a result of nitrate reduction rather than ammonia oxidation (Einsele and Wetter, 1938).

The excretion of extracellular nitrite by phytoplankton as observed by Orr (1926) and Zobell (1935) may also influence the distribution of nitrite within the surface layers of natural waters (Rakestraw, 1936; Hutchinson, 1957; Vaccaro and Ryther, 1960).

Nitrate in Sabkhet El-Bardawil waters showed variations during the study period. The seasonal average values for the surface water ranged between 39.58 - 155.21 ug/l and 35.40 - 164.58 ug/l near the bottom. The annual average values were 70.73 and 72.36 ug/l for surface and bottom layers respectively (Fig. 2).

Nitrate increases slightly downwards. The concentration of nitrate in the water column is governed by the advective transport of nitrate into surface layers, the microbial oxidation of ammonia and the uptake by the primary producers. The light penetration is sufficient to reach the bottom all the year-round due to the general shallowness of the Lagoon. The uptake is usually much faster than the processes transporting nitrate into the surface layers. Therefore, the nitrate concentration in surface water is low. This was indicated also by Grasshoff (1976) and stated that the concentration of nitrate in most surface waters is close zero.

The relatively high values of nitrate recorded in winter 1986 may be due to the decomposition of dead algae (Zobell, 1946; Zafar, 1964; Siliem, 1974). Fig. 2 shows a nitrate maximum concentration in spring and autumn 1986 and minimum concentration in summer. A similar trend was observed in Lake Manzalah (El-Wakeel and Wahby, 1970); Lake Maryut (Wahby, 1961, & Samaan, 1966 and for Lake Quarun (Naguib, 1958), also in the sea (Vaccaro, 1965).

In the clean waters, nitrate concentration varies from 4.1 to 602 ug/l (Sverdrup et al., 1955). The values obtained for the Lagoon is clearly less than those values recorded for the Oceans.

In Bardawil Lagoon the surface average phosphate ranged between 9.69 ug/l in early autumn 1986 to 425.98 ug/l in late autumn 1985. The bottom seasonal average concentration is higher, especially over the bottom which ranges between 3.55 ug/l in October, 1986 to 462.13 ug/l in November, 1985. While the regional average values were 152.11 and 158.78 ug/l for surface and bottom water layers respectively (Fig. 3).

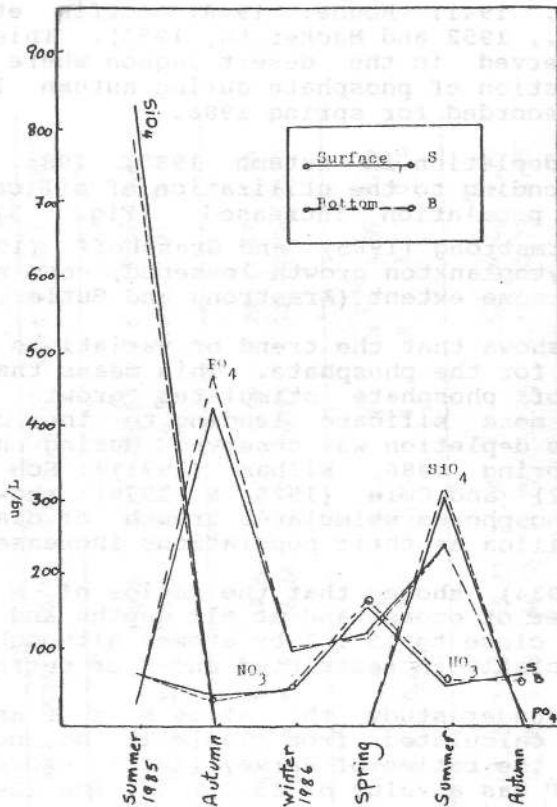


FIG. 3
Phosphate Nitrate and Silicate Variations
during 1985 and 1986

The phosphorus content of seawater has been estimated to be about 61.96 ug/l, but in euphotic zone, the level was usually much lower (Grasshoff, 1976). The values obtained for the Bardawil Lagoon being higher than that obtained by Grasshoff due to the arid climate. Hutchinson (1957) listed Good Enough Lake, British Columbia, with 208 g/m³ which is 10,000 times the mean from dilute lakes in humid regions.

In lakes, phosphorus is affected by both utilization by phytoplankton and the interchange between mud and the overlying water. Phosphate concentration may limit the rate of photosynthesis (Goldman, 1961). Phytoplankton not only utilize this trace of phosphorus for their growth and reproduction but also they store phosphorus in their tissues

in big quantities if phosphorus in water allow it (Ruttner, 1968; Einsele, 1941; Rodhe, 1948; Coffin et al., 1949; McCarter et al., 1952 and Mackereth, 1953). This phenomenon is clearly observed in the desert lagoon where there is a complete exhaustion of phosphate during autumn 1986 and the lower values recorded for spring 1986.

A marked depletion in autumn 1985, 1986, winter and spring corresponding to the utilization of silicates as the phytoplankton population increased (Fig. 3). This is confirmed by Armstrong (1965) and Grasshoff (1976). During the summer phytoplankton growth lessened, and silicate was regenerated to some extent (Armstrong and Butler, 1960).

Figure. 3 shows that the trend of variations of silicate is the reverse for the phosphate. This means that, the high concentration of phosphate stimulates growth of diatoms, which utilize more silicate leading to the depletion of silicate. This depletion was observed during autumn 1985, winter and spring 1986. Kilham (1971); Sch elske and Steormer (1972) and Cole (1975 & 1979) showed that the addition of phosphorus stimulates growth of diatoms; which utilize more silica as their populations increase.

Redfield (1934) showed that the ratios of N : P in the sea, in a number of oceans and at all depths and in plankton were generally close to 15 : 1 by atoms; although exceptions do exist, especially in restricted and / or surface water.

In the area under study the ratios N : P and $\text{NO}_3 : \text{PO}_4$ (Table 2 were calculated from Table 1) do not show any agreement with the ratios of Harvey (1926) except in autumn 1986 where N:P has a value of 15.5 : 1. The low ratio of N : P (3:1), for both surface and bottom layers, recorded in spring and the ratio of N : P (6 & 4:1) for both surface and bottom respectively recorded in summer 1985, indicate a faster assimilation of nitrate than for phosphate. The very lower ratios recorded were in agreement with the observations of Ketchum et al., 1958 who found the ratio in English coastal surface waters varied from zero in summer to values of 5:1 and 10:1 in winter. On the continental shelf they were from 1.2:1 to 7:1 (Armstrong, 1965). On the other hand, the high ratio Si:P (35:1) and (23:1), for both surface and subsurface respectively, indicate a stacking of diatoms and regeneration of silicon. El-Rayis (1973) obtained the ratio of Si:P of high value (17-55 : 1) in Alexandria water which nearly closes to the ratio obtained for the lagoon waters.

SUMMARY

The concentrations of nutrient salts showed regional and seasonal variations. Ammonia values are slightly less than sea water values, and increased downwards. Nitrite showed with some exceptions a complete depletion and its trend is correlated to nitrate variations. Nitrate increases slightly downwards and is governed by advective transport of nitrate into surface layers, the microbial oxidation of ammonia and the uptake by the primary producers.

Table 1
Seasonal average concentrations of nutrient salts (ug/l & ug at/l) in Bardawil Lagoon.

	1985						1986					
	Summer		Autumn		Winter		Spring		Summer		Autumn	
	ug/l	ugat/l	ug/l	ugat/l	ug/l	ugat/l	ug/l	ugat/l	ug/l	ugat/l	ug/l	ugat/l
3+												
PO ₄	S	5.95	0.84	425.98	13.75	95.98	3.1	119.8	3.87	235.58	7.6	9.69
	B	37.95	1.22	462.13	14.92	97.54	3.15	114.8	3.71	114.8	7.63	3.55
+												
NH ₃	S	22.73	1.62	20.5	1.46	17.08	1.36	19.64	1.40	24.99	1.79	41.48
	B	21.02	1.50	18.08	1.29	20.67	1.48	16.64	1.19	20.95	1.50	37.76
NO ₂	S	N	N	N	N	7.0	0.5	15.75	1.12	N	N	4.09
	B	N	N	N	N	6.4	0.45	11.9	0.85	N	N	3.65
NO ₃	S	68.75	4.91	39.58	2.83	43.75	3.13	155.21	11.09	50	3.57	67.08
	B	67.05	4.79	35.42	2.53	46.88	3.35	164.58	11.75	57.95	4.14	62.29
SiO ₄	S	831.21	29.62	N	N	N	N	N	N	300.05	10.7	N
	B	776.99	27.69	N	N	N	N	N	N	310.18	11.05	N

S = Surface
B = Bottom

Table 2
 Silicon: Nitrogen: Phosphorus Ratios (by atoms and by ug/l) in the Bardawil Lagoon

		S i l i c o n		N i t r o g e n		P h o s p h o r u s					
		B		S		B					
		ug/l	ugat/l	ug/l	ugat/l	ug/l	ugat/l				
		ugat/l	ugat/l	ugat/l	ugat/l	ugat/l	ugat/l				
Summer 1985	32	3	20.5	23	2.65	6	1.77	4	1	1	1
Autumn	N	N	N	N	0.09	0.21	0.08	0.17	1	1	1
Winter 1986	N	N	N	N	0.46	1.0	0.48	1.1	1	1	1
Spring	N	N	N	N	1.3	3	1.4	3	1	1	1
Summer	1.27	1.4	1.31	1.5	0.21	0.5	0.25	0.5	1	1	1
Autumn	N	N	N	N	6.9	15.5	17.5	40	1	1	1

S = Surface

B = Bottom

Phosphate concentration in Bardawil Lagoon is higher than that recorded for sea water due to the arid climate. In additions, phosphate showed a complete exhaustion during autumn and the lower values were recorded in spring.

A marked depletion of silicate was observed in autumn, winter and spring corresponding to its utilization of silicate. The trend of silicate variation is the reverse of that for phosphate.

The low N:P ration indicates a faster assimilation of nitrate than phosphate, while the high Si:P ratio indicates a lessening of diatoms and regeneration of silicates.

ACKNOWLEDGMENT

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REFERENCES

- Anon., 1969. Standard methods for examination of water and waste water. 12 th Ed. New York.
- Armstrong, A.J., 1965. Silicon. In: Chemical Oceanography (J.P. Riely and S. Skirrow, eds). Academic Press, London, part I pp: 409-432.
- Armstrong, A.J. and E.I. Butler., 1960. Chemical changes in Sea water off Plymouth During 1958. J. Mar. Biol. Ass. U.K., 39: 299-302.
- Coffin, C.C.; F.R. Hayes; L. Juday and S.G. Whiteway. 1949. Exchange of materials in a lake as studied by the addition of radioactive phosphorus. Canada. J. Res., D. 26.
- Cole, G.A., 1975. Text book of limnology. Saint Louis, The C.V. Mosby company, 283 P.
- Cole, G.A 1979. Text book of limnology. St. Louis The C.V. Mosby Company 426 p.
- Einsele, D.W., 1941. Die Umsetzung von Zuge fuhrtem an organischem Phosphate in autropen sea und ihre Rukivinkungen auf sziner Gesamthaushalt. Zacher. Fisch., pp. 39.
- Einsele, W. and H. Wetter, 1938. Untersuchungen Under die Entwicklung der Ptsikalischen und Chemischen Verhaltnisse in Jahrezklus in einen massing eutrophen see (Schleinsee bei Langenargen). Int. Rev. Hydrobiol., 36: 285-324.
- El-Rayis, O.A., 1973. Cycle of nutrient salts in Mediterranean Sea water of Alexandria region. M.Sc. Thesis, Alex. Univ., 150 P.
- El-Wakeel, S.K. and S.D. Wahby, 1970. Hydrography and Chemistry of Lake Manzalah, Egypt. Arch. Hydrobiol., 67: 173-200.
- Goldman, C.R., 1961. Verh. Int. Ver. Limnol., 24: 120 In: J.D.H. Strickland. Organic productivity. Chemical Oceanography, Vol. I (1965) London, 477-610.
- Grasshoff, K., 1976. Methods of Sea water Analysis. Verlag. Chemie. Weinheim. New York, 317 P.
- Harvey, H.W., 1926. Nitrate in the Sea. J. Mar. Biol. Ass. U.K., 14: 71-88.
- Hutchinson, G.E., 1957. A treatise on limnology. Vol. I, John Wiley and Sons, Ltd., New York, 1015 P.

- Ketchum, B.H., R. F. Vaccaro, and N. Corwin, 1958. The annual cycle of phosphorus and nitrogen in New England coastal waters. *J. Mar. Res.*, 17: 282-301.
- Kilham, P., 1971. A hypothesis concerning silica and the freshwater planktonic diatoms. *Limnol. Oceanogr.*, 16: 10-18.
- Mackereth, F.J., 1953. Phosphorus utilization by *Asterionella formosa*. *J. Exptl. Bot.* 4.
- McCarter, J.A.; F.R. Hayes; H.L. Jordan and M.K. Cameron, 1952. Movement of materials in the hydrolimnion of lakes as studied by addition of radioactive Phosphorus. *Canad. J. Zool.*, 30.
- Naguib, M., 1958. Studies on the ecology of Lake Quarun. Part. 1. Kieler Meeresforschungen, XIV, H. 2.
- Orr, A.P., 1962. The nitrate content of sea water. *J. Mar. Biol. Ass., U.K.*, 14: 55-61.
- Rakestraw, N.W., 1963. The occurrence and significance of nitrite in the sea. *Biol. Bull.*, 71.
- Redfield, A.C., 1934. On the proportions of organic derivatives in sea water and their relation to the composition of plankton. p.176-92. In: James Johnstone Memorial Vol., Liverpool, Univ. Press, 348 P.
- Rodhe, W., 1948. Environmental requirements of freshwater plankton algae. *Symb. Bot.*, Upsala, X: 1. In: J. D. H. Strickland. Organic Productivity. *Chemical Oceanography*, Vol I (1965) London. 477-610. P.
- Ruttner, F., 1968. *Fundamentals of Limnology*. University of Toronto press. 295 P.
- Samaan, A.A., 1966. Primary production in Lake Maryut. Ph. D. Thesis, Alex. Univ.
- Schelske, C.L. and E. F. Stoermer, 1972. Phosphorus, silica, and eutrophication of Lake Michigan. pp. 157-171. In: G. E. Likens, ed., *Nutrients and eutrophication: the limiting-nutrient controversy*. ASLO Special Symposia 1. American Society of Limnology and Oceanography. Lawrence, Kansas.
- Siliem, T.A.E., 1974. The chemical changes of the water of the Manzalah Fish Ponds. M.Sc. Thesis, Fac. Sc., Alex. Univ., 106 P.
- Siliem, T.A.E., 1984. Chemical studies on pollution in the Damietta Nile Branch between the Faraskour Dam and Ras El-Bar Outlet. PH. D. Thesis, Fac. Sc. Alex. Univ. 253 P.
- Siliem, T.A.E., 1989. Chemical Conditions in Bardawil Lagoon. 1-The Major Cations. *Bull. Nat. Instit. Oceanogr. & Fish.*, (this volume).
- Siliem, T.A.E., 1989. Chemical Conditions in Bardawil Lagoon. 2-The Major Anions. *Bull. Nat. Instit. Oceanogr. & Fish.*, (this volume).
- Sverdrup, H.U.; M. W. Johnson and R. H. Fleming, 1942. *The Oceans*. Prentice Hall, New York. 1087 P.
- Sverdrup, H.U.; M. W. Johnson and R. H. Fleming, 1955. *The Oceans*, Prentice Hall, New York, 1087 P.
- Vaccaro, R.F., 1965. Inorganic Nitrogen in Seawater. *Chemical Oceanography Vol. I*: 365-408.
- Wahby, S. D., 1961. Chemistry of Lake Marut. Note and Mem. No. 65, Alex. Instit. of Oceanog. & Fish.
- Yitzhak, L., 1971. Anomalies of Ca and SO₄ in the Bardawil Lagoon Northern Sinai. *Limnol. Oceanogr.*, 16 (6): 983-987.
- Zafar, A.R., 1964. On the ecology of algae in certain fish ponds of Hyderabad, India. I. Physico. Chemical Complexes. *Hydrobil.*, 23 (1-2): 179-196.

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Zobell, C.E., 1935. The assimilation of ammonium nitrogen by *Nitzschia closterium* and other marine Phytoplankton. *Nat. Acad. Sci., Proc.*, 21: 517-522.

Zobell, C.E., 1946. Marine Microbiology. *Chronica Botanica Waltham, Massachusetts, U. S. A.* In: Ralf F. Vaccaro, Inorganic Nitrogen In Sea water. *Chemical Oceanography, Vol. I:* 365-408.

The water in the lagoon... (faint text describing water characteristics and sampling methods)

DISCUSSION

Most of the water samples taken... (faint text discussing the significance of the water samples and the lagoon environment)

In this study, the water samples were taken at various depths... (faint text detailing the sampling procedure and the locations of the samples)

The data obtained from this investigation is... (faint text summarizing the findings and their importance in understanding the lagoon's chemistry)

REFERENCES

Reports of the... (faint text listing references related to the study, including mentions of tables and figures)